

**AMENDMENTS TO THE SPECIFICATION:**

**On page 1, please delete the second complete paragraph and insert the following replacement paragraph:**

-- As the cationically polymerizable organic substances, many compounds have been well known such as methylol compounds, ethylene compounds, polyacetal compounds, organosiloxane compounds, polyamide compounds, heterocyclic compounds, etc. --

**On page 1, please delete the fourth complete paragraph and insert the following replacement paragraph:**

-- Cationic polymerization of epoxy compound or organosiloxane ~~compoundcompounds~~, which is a means for producing epoxy resin or silicone resin, is carried out in the presence of polymerization catalyst or polymerization initiator. --

**On page 1, please delete the sixth complete paragraph and insert the following replacement paragraph:**

-- Regarding photopolymerization of epoxy-~~compoundcompounds~~, the example of initiator for cationic ~~photopolymerizationinphotopolymerization~~ is onium salt which contains an element having a lone pair to which a proton or other cation compound is bonded via a coordinate bond. Typical ~~exampleexamples~~ of the onium salt ~~isare~~ aromatic diazonium salt, aromatic iodonium salt, and aromatic sulfonium salt. Many of the onium salts contain halogen metal complex anion ( $\text{BF}_4^-$ ,  $\text{BF}_6^-$ ,  $\text{AsF}_6^-$ ,  $\text{Sb F}_6^-$ , etc.) as counter ion. --

**On page 2, please delete the first complete paragraph and insert the following replacement paragraph:**

-- The working mechanism of these conventional photopolymerization initiators such as diazonium salt, iodium salt, and sulfonium salt is as follows: At first, Brønsted acid is produced by U.V. light irradiation. Then, the produced Brønsted acid reacts with the cationically polymerizable organic substances and the molecular chair of the polymer grows. --

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**On page 2, please delete the third complete paragraph and insert the following replacement paragraph:**

-- Photopolymerization initiator containing borate counter anion has been disclosed in Japanese Patent Laid-open Nos. 143,044/87 and 182,701/90. In accordance with these literatures, the working mechanism is as follows: When dyestuff which is cation component of the complex is exposed to U.V. light, it is excited to singlet state; it receives an electron from the borate salt which is anion component; then the produced borate radical dissociates one of the ligands to generate a radical, whereby the radical polymerization proceeds. --